

Past, present and future of hematite (and other metal oxide photoanodes) for solar water splitting



Dr. Camilo A. Mesa

cam111@ic.ac.uk

 Durrant
Group

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BREWDOG TOMORROW

LET'S ENSURE WE HAVE A PLANET TO BREW BEER ON.

BREWDOG HAS COME A LONG WAY.

WE'VE GROWN, AND WE'VE GROWN UP.

WE HAVE ALWAYS BELIEVED THAT BUSINESS SHOULD BE A FORCE FOR GOOD AND THAT BRAVE THINKING AND BOLD ACTIONS ARE THE ONLY WAY TO MAKE REAL IMPACT.

TODAY, WE ARE IN THE MIDDLE OF A CLIMATE CRISIS. IT IS A CRISIS OF OUR OWN DESIGN, DRIVEN BY BIG BUSINESS. WE NEED TO RECONSIDER OUR VALUES AND THE IMPACTS OF OUR ACTIONS. WE DON'T HAVE ALL THE ANSWERS, BUT IN 2020 WE COMMITTED TO MAKING GREAT BEER, AND WE HAVE A PLANET TO DRINK IT ON.

NOW IS THE TIME TO BE RADICAL IN EVERYTHING WE DO.

WE BELIEVE THE BEST SOLUTIONS WILL COME THROUGH TRANSPARENCY, COLLABORATION AND COMMITMENT. IT'S THE ONLY WAY TO ACCELERATE THE CHANGE WE NEED.

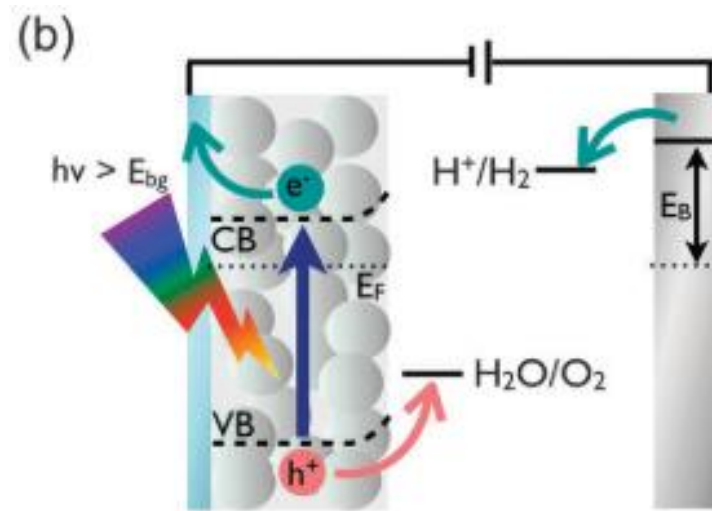
ACTIONS NOT PROMISES.

6CO₂ + 6H₂O → C₆H₁₂O₆ + 6O₂

THE NEW BREWDOG PHOTO HERE

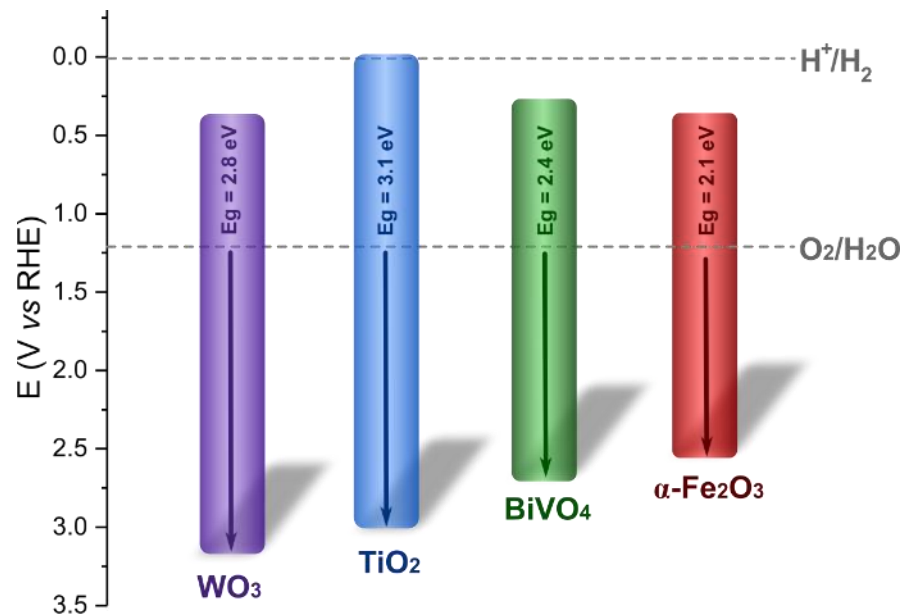
CHANGE IS BREWING.

Solar Fuels



Photoelectrochemical water splitting

Cowan, A. and Durrant, Chem Soc Rev., 42, 2281, 2013

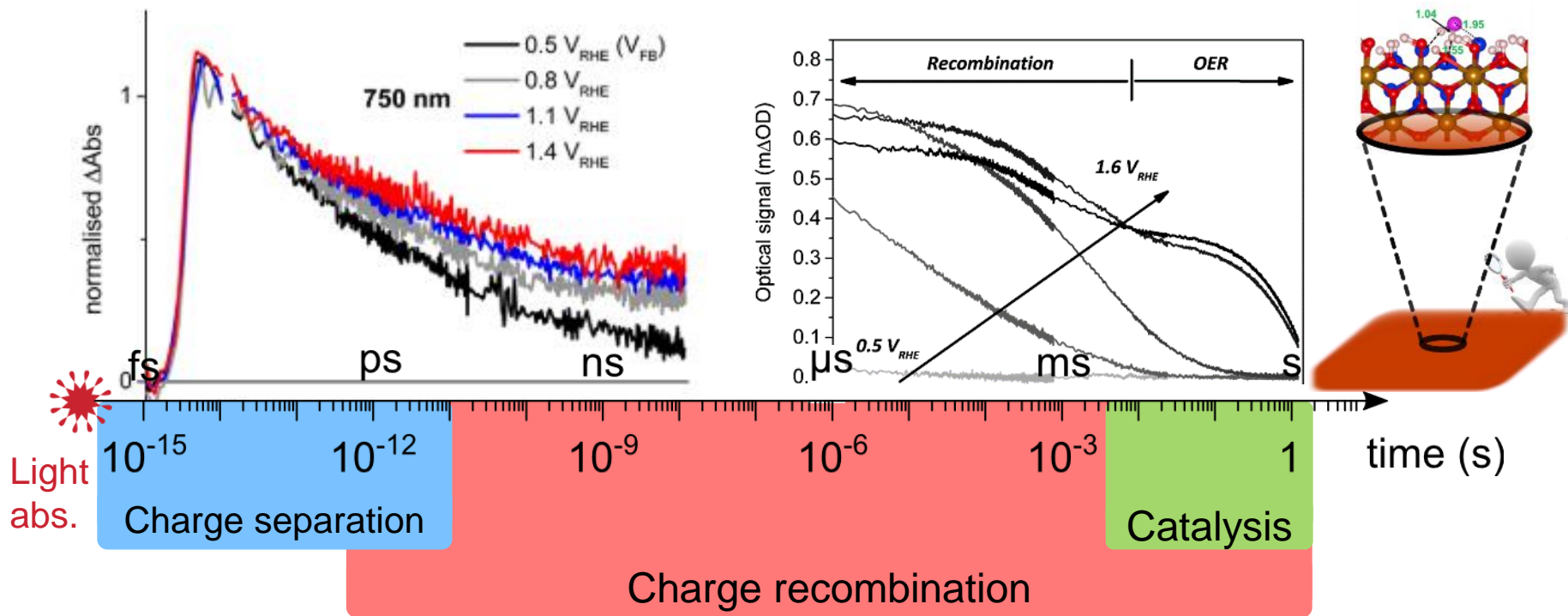


Photoelectrochemical water splitting

Corby, S., et. al., JACS, 140, 16168, 2018

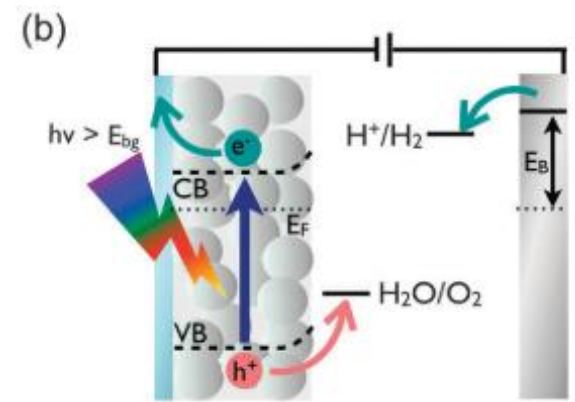
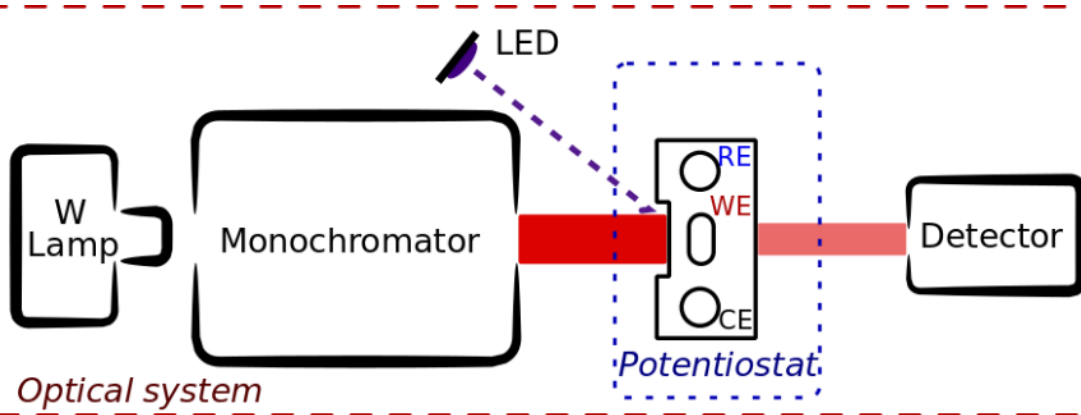
The lifetime challenge

Example of $\alpha\text{-Fe}_2\text{O}_3$ photoanodes

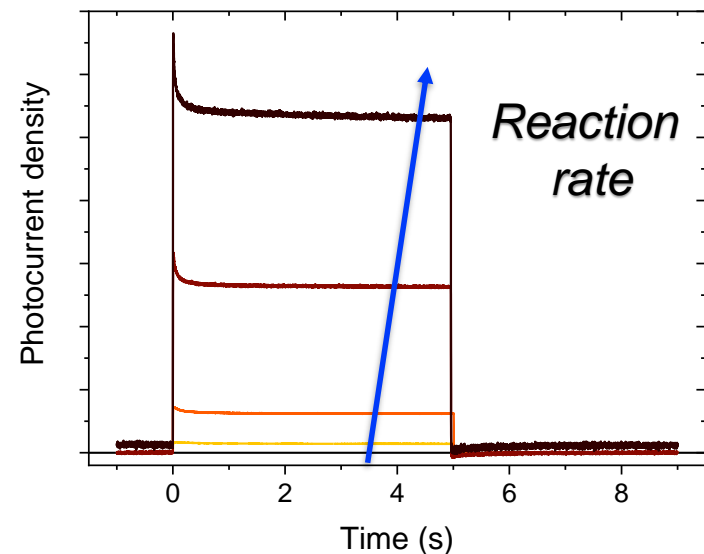
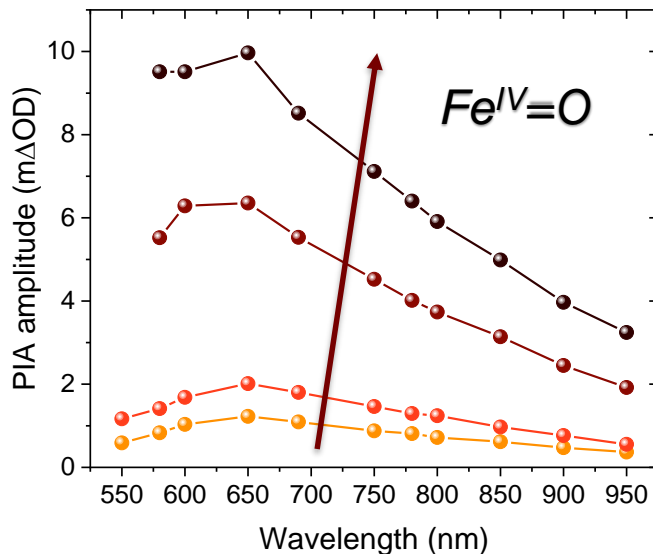


Pendlebury, S., et al., *JACS.*, 136, 9854, 2014

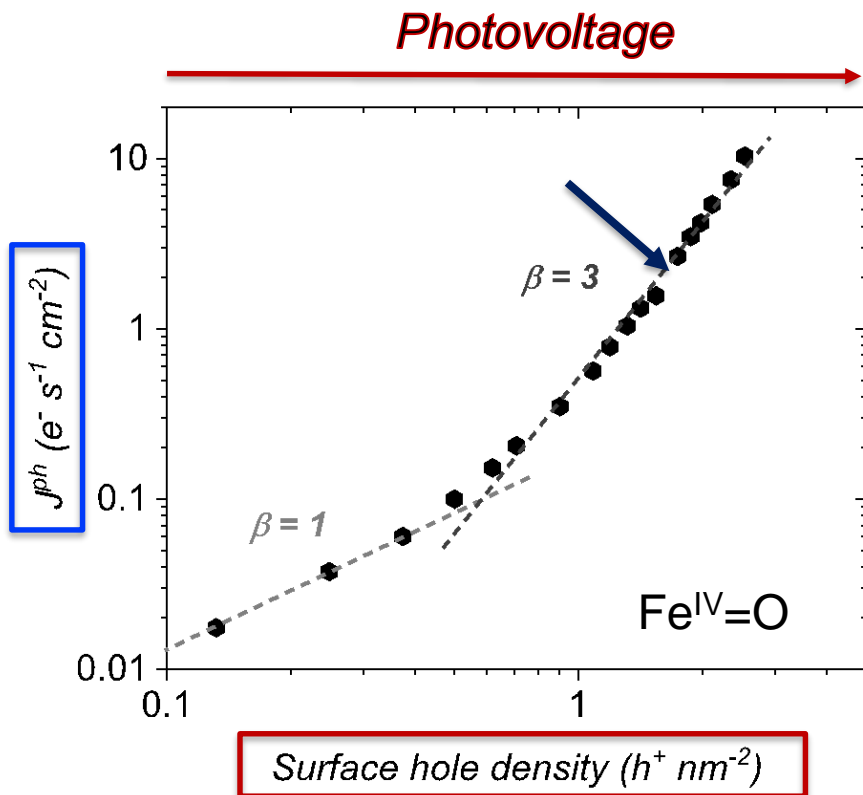
Francàs, L., Mesa, C., et. al., *Rate law analysis in water splitting photoanodes in: Advances in photoelectrochemical water splitting, RSC, 2018*



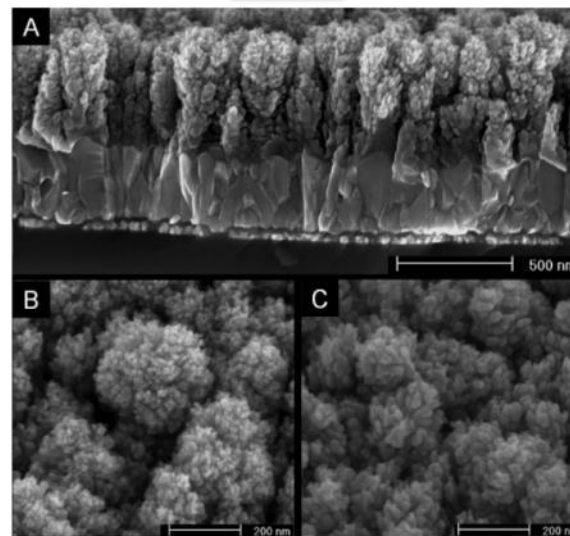
Increasing light intensity



$$\log(J^{ph}) = \beta \cdot \log(h_s^+) + \log(k_{WO}) \quad \leftarrow \quad J^{ph} = k_{WO} \cdot h_s^{\beta}$$

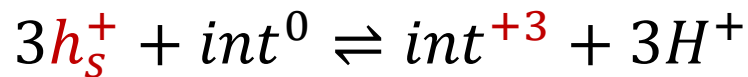


Le Formal, F. et al., *J Am Chem Soc*, 2015, 6629

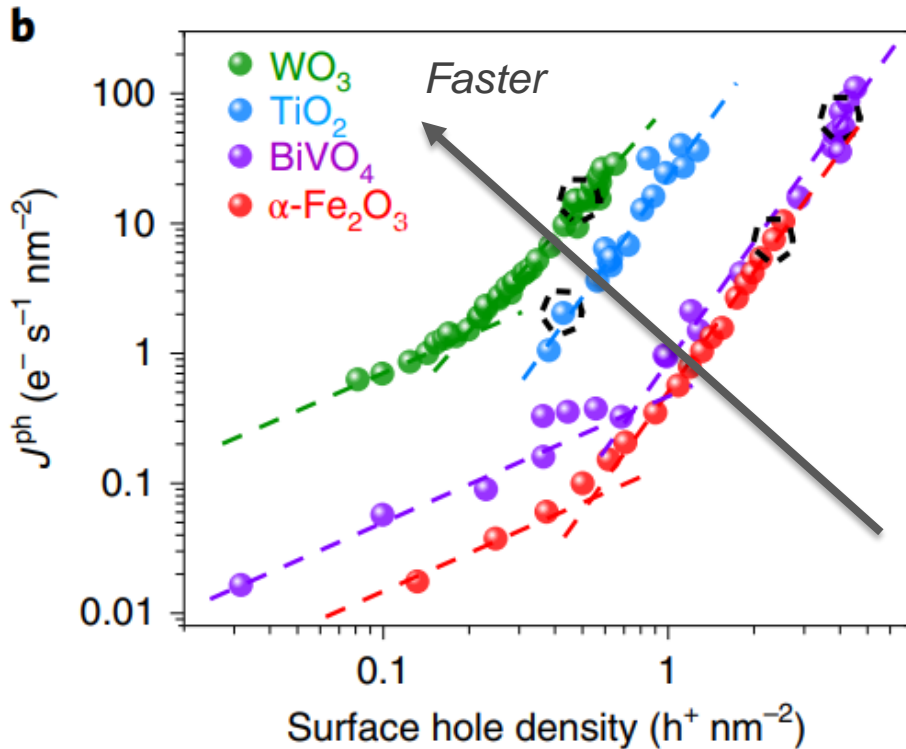


Kay A. et al., *JACS*, 128, 15714, 2006

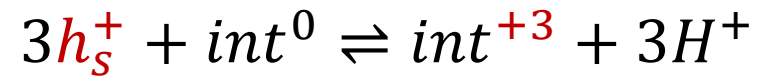
Molecular approach



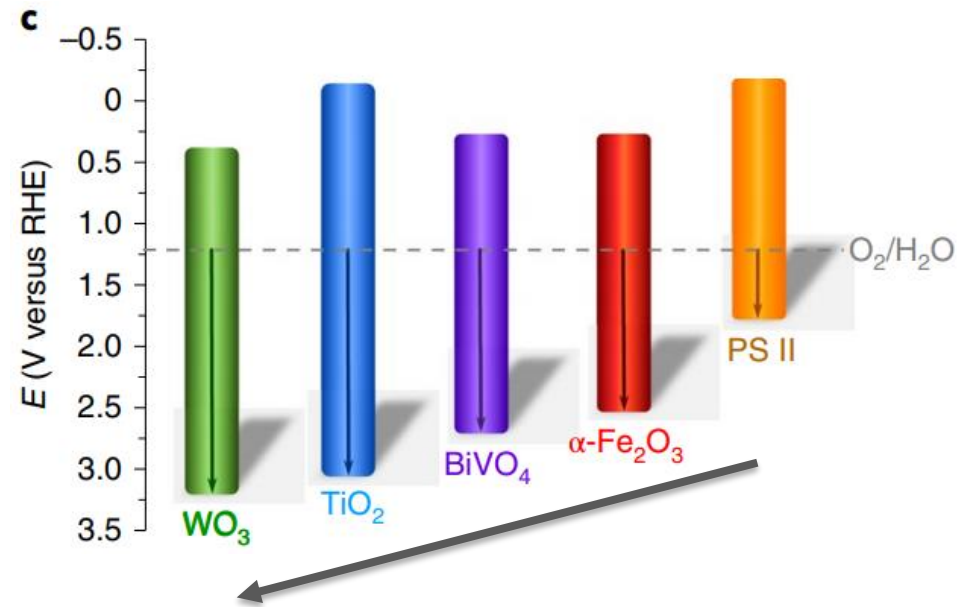
$$r = k_{WO} \cdot (h_s^+)^3$$



Molecular approach



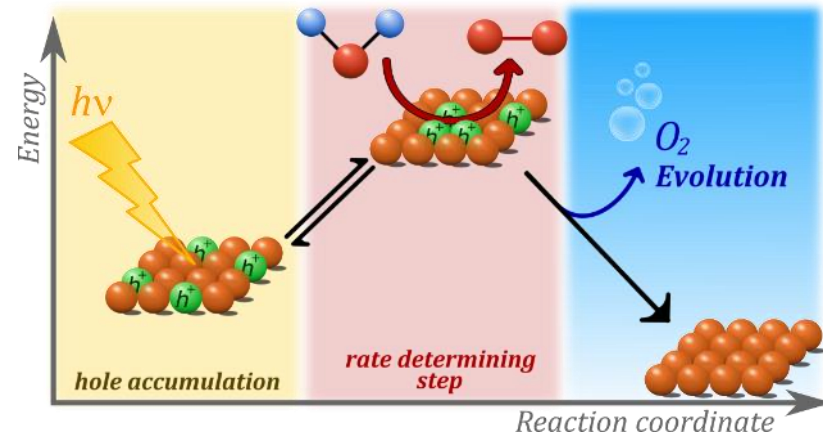
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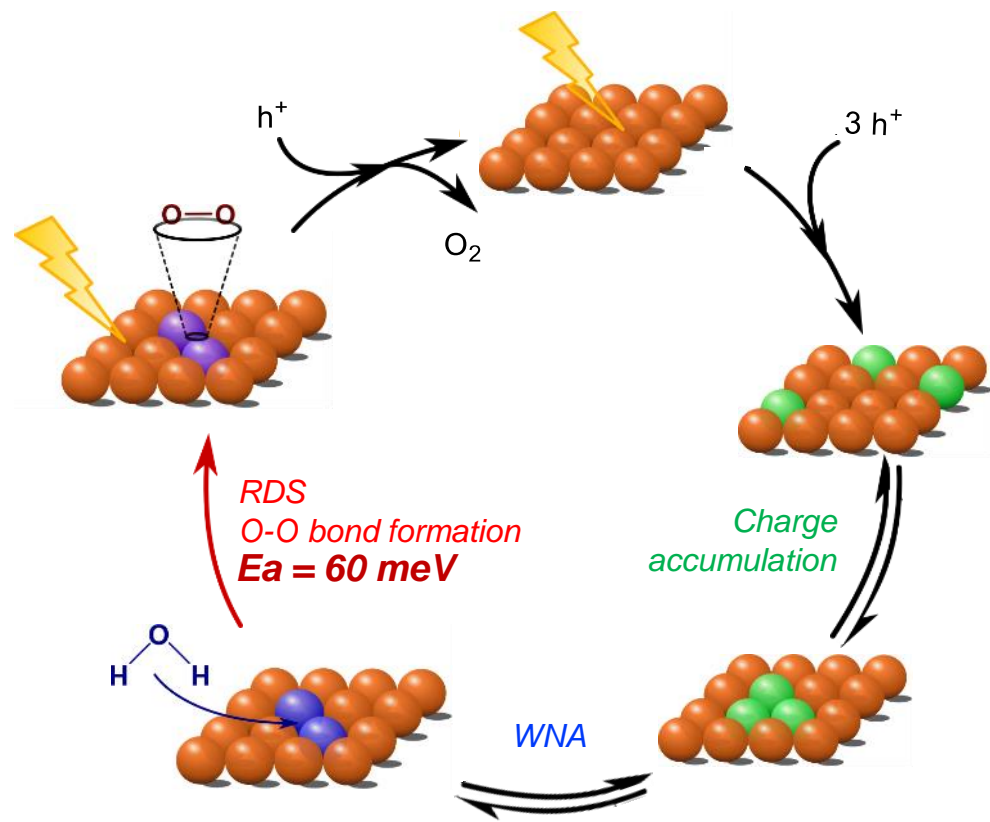
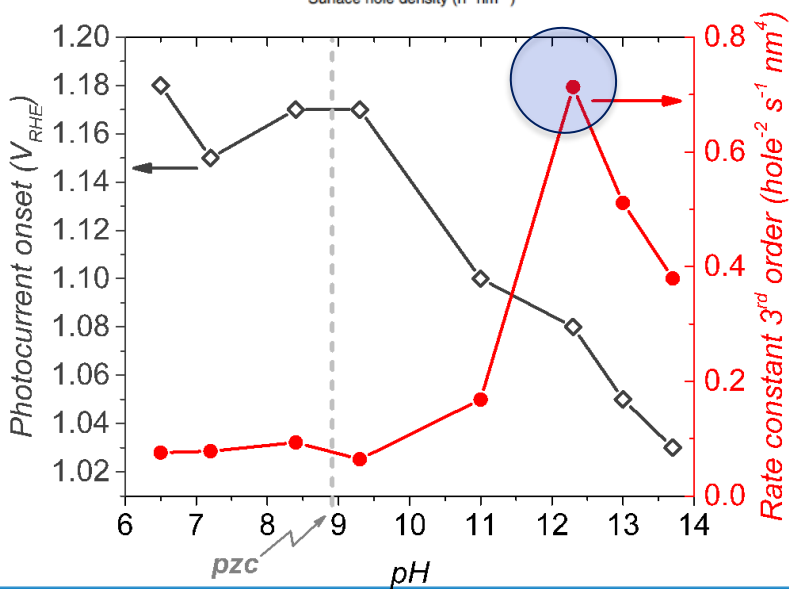
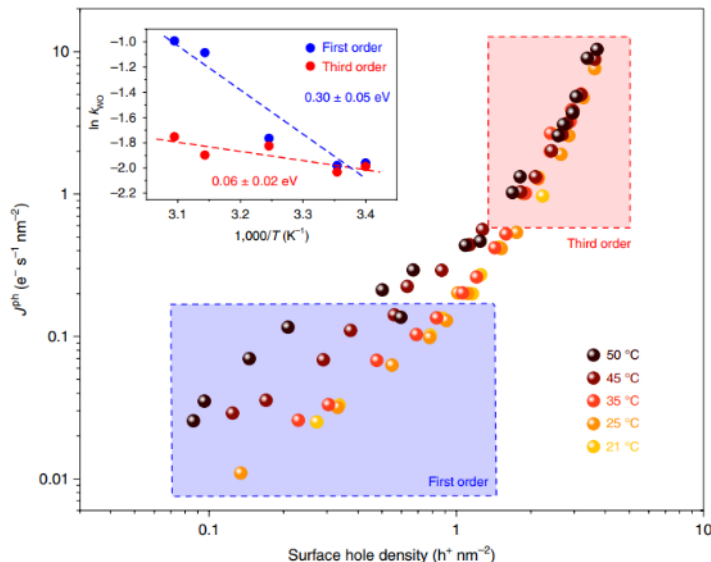


Mesa, C., Francàs, L., et al., Nat. Chem., 12, 82, 2020

First concluding remark

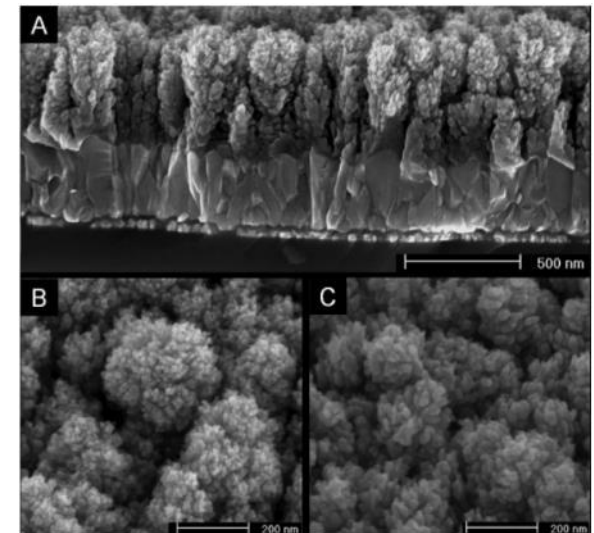
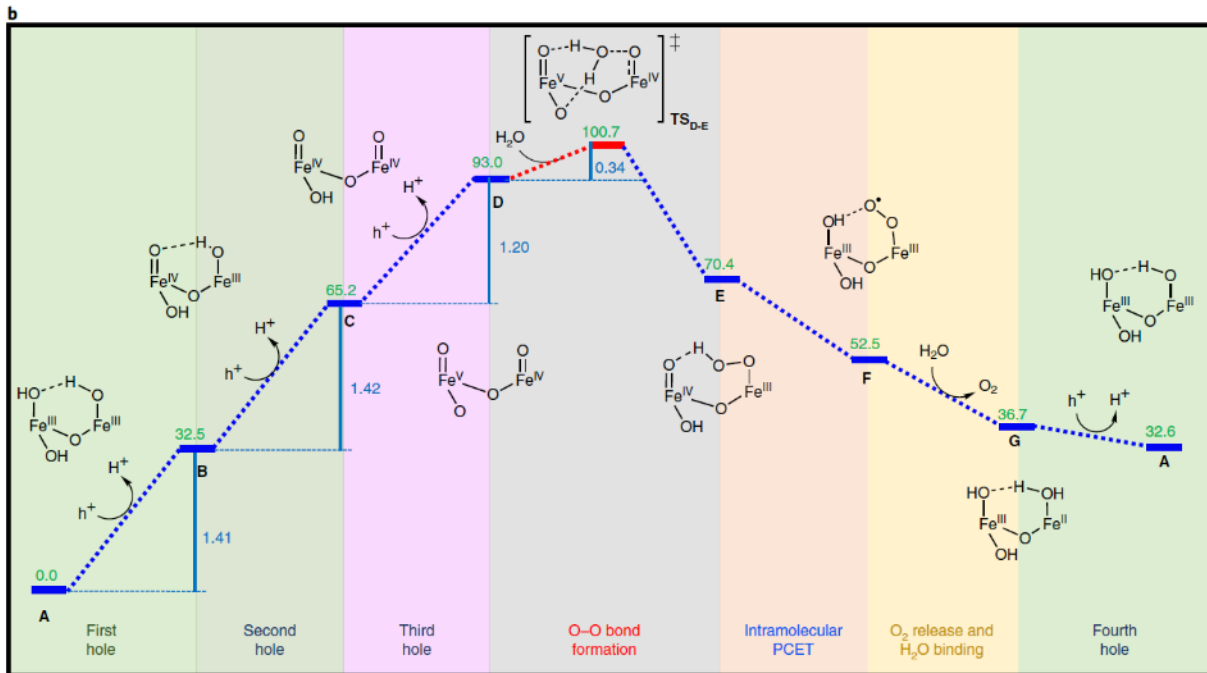
The studied materials ($\alpha\text{-Fe}_2\text{O}_3$, TiO_2 , BiVO_4 and WO_3) share common oxidative pathways of reaction, based on accumulation of charges, with kinetic differences only associated to the hole redox power





Mesa, C., Francàs, L., et al., Nat. Chem., 12, 82, 2020

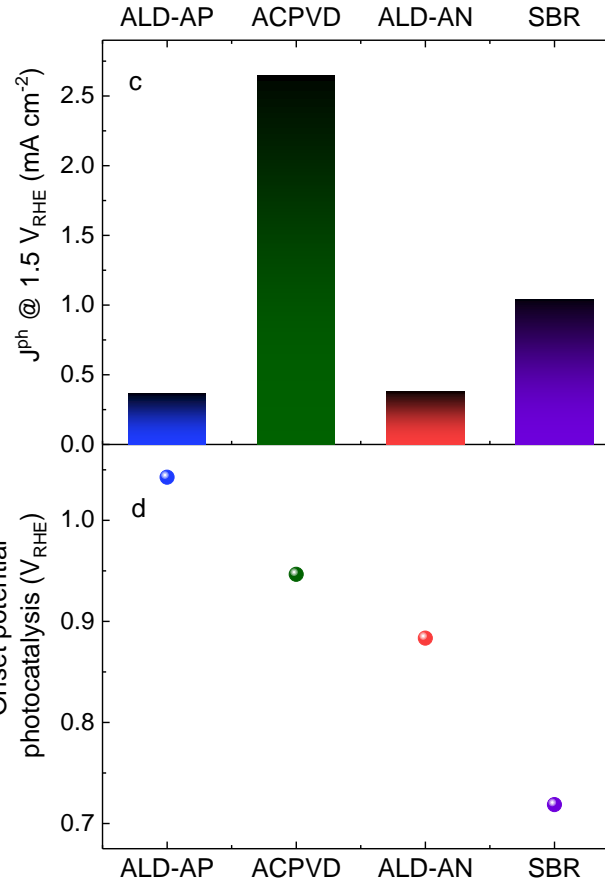
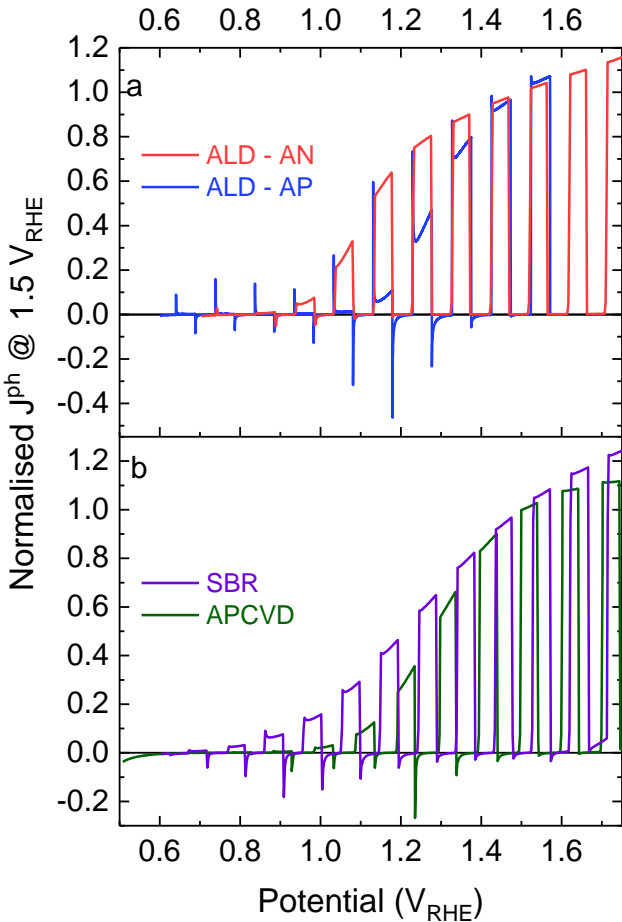
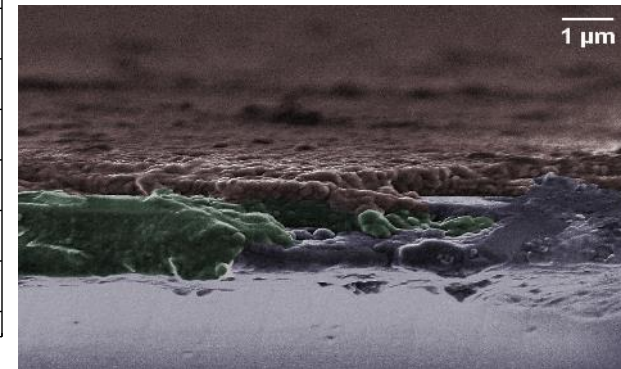
Morphology effect?



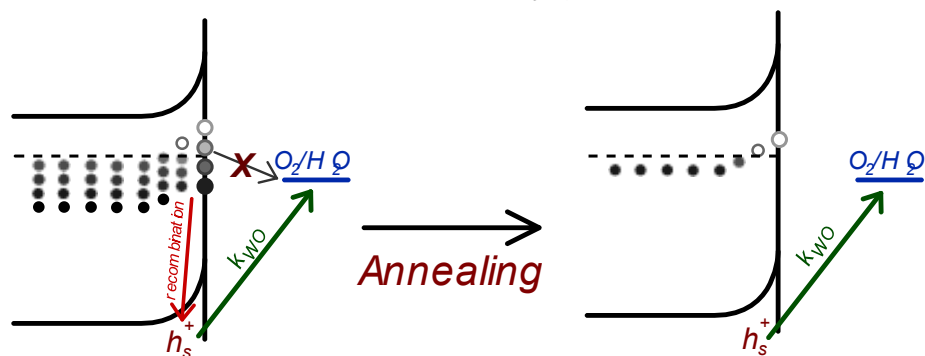
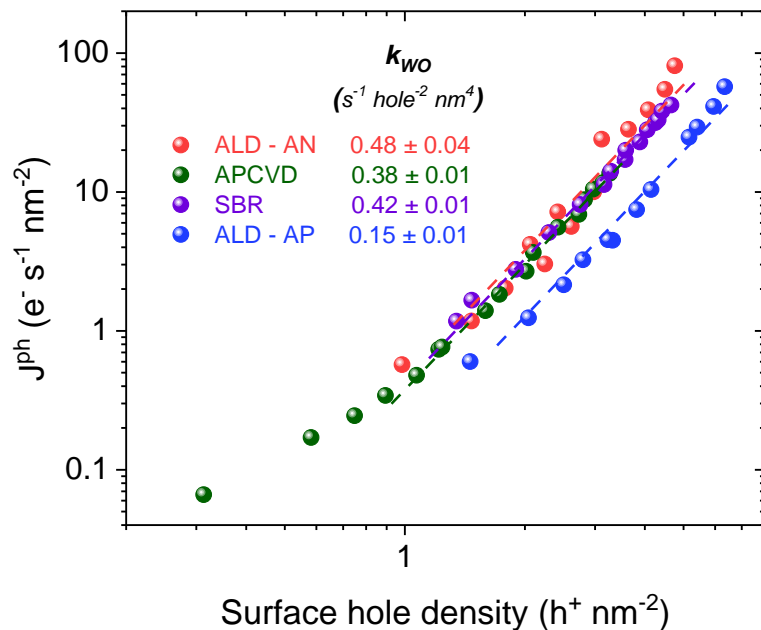
Mesa, C., Francàs, L., et al., *Nat. Chem.*, 12, 82, 2020



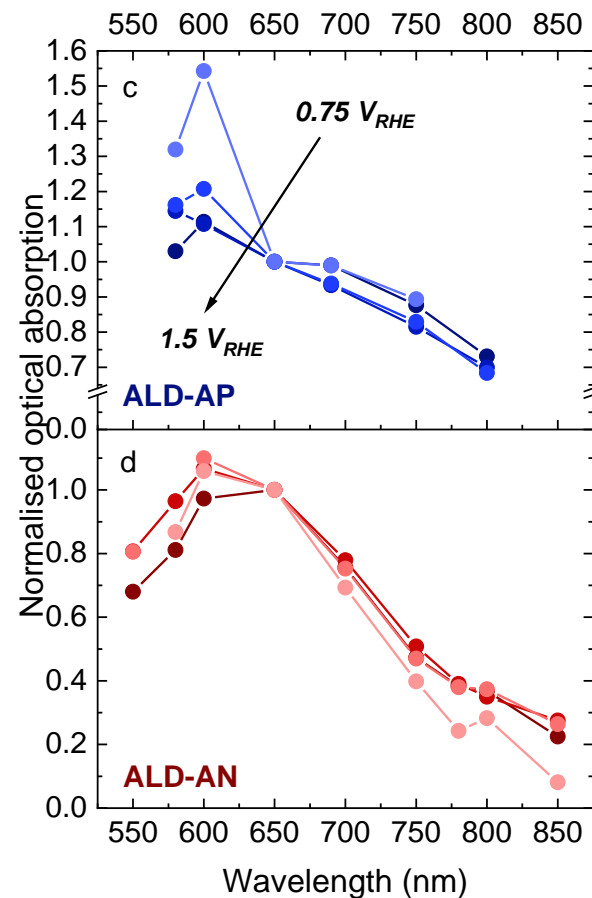
Steier, L. et al., ACS Nano, 9, 11775, 2015



Mesa, C., et al., Manuscript in preparation



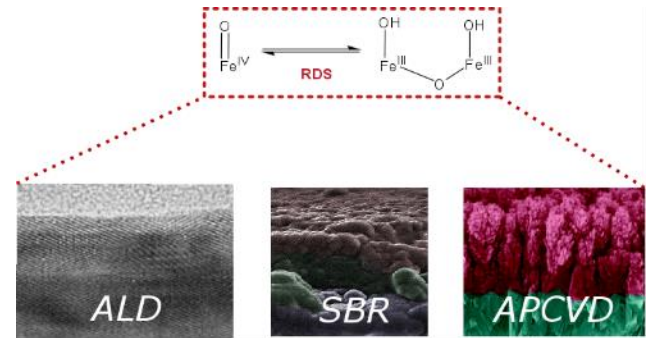
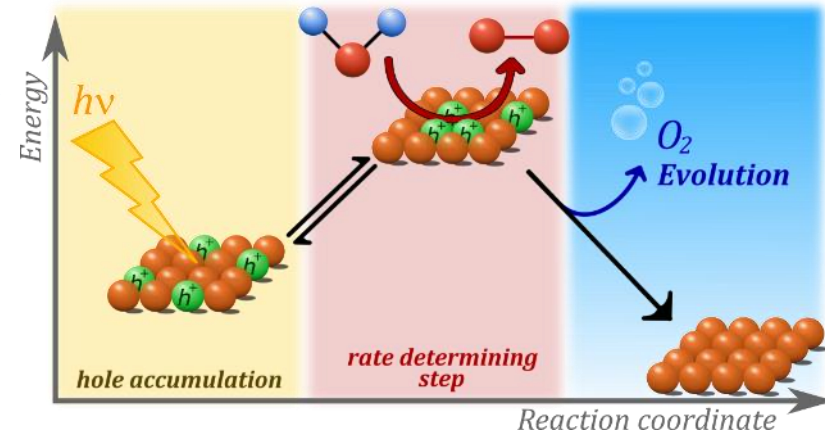
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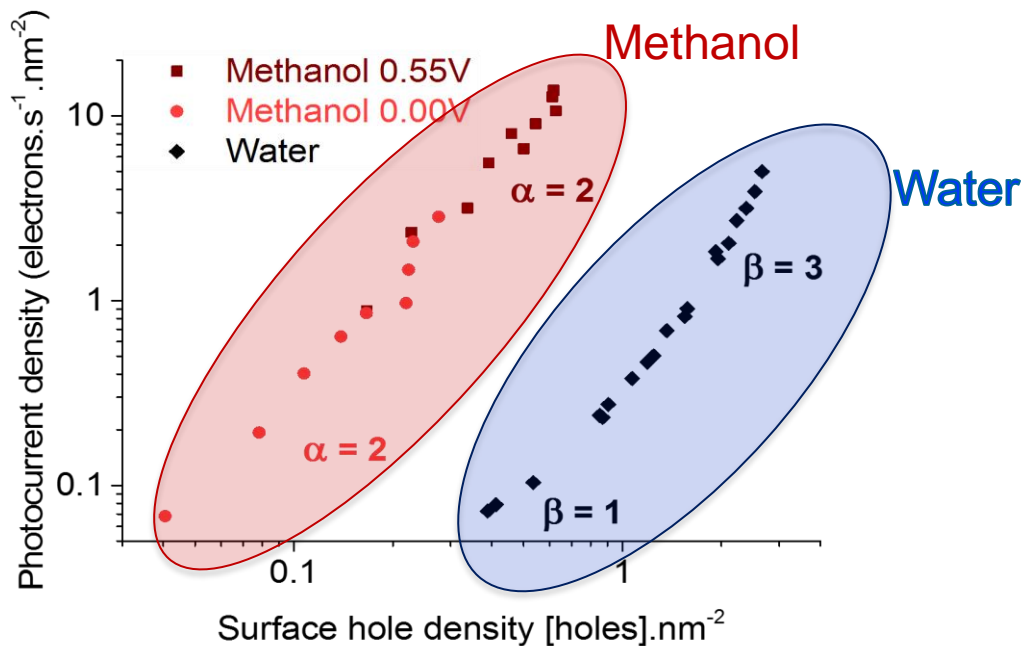


Second concluding remark

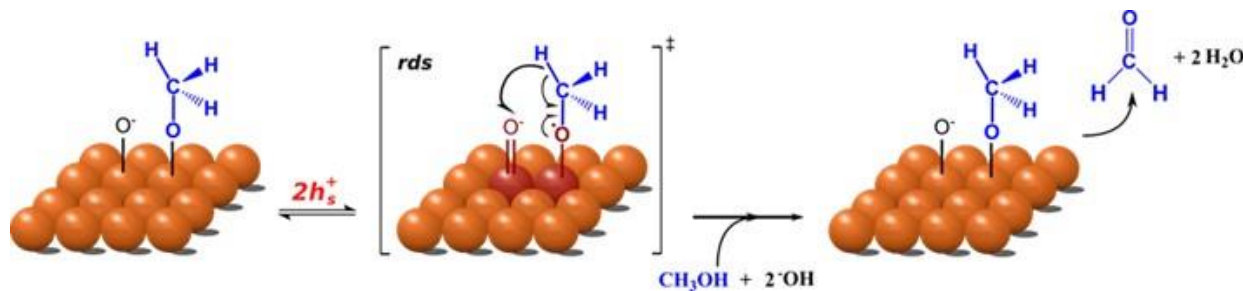
The studied materials ($\alpha\text{-Fe}_2\text{O}_3$, TiO_2 , BiVO_4 and WO_3) share common oxidative pathways of reaction, based on accumulation of charges, with kinetic differences only associated to the hole redox power

Differently synthesized hematite photoanodes perform water oxidation with the same mechanism





*Proof of concept
for oxidation of
alcohols with
 $FE \sim 1$*



Mesa, C. et. al., JACS, 139, 33, 2017

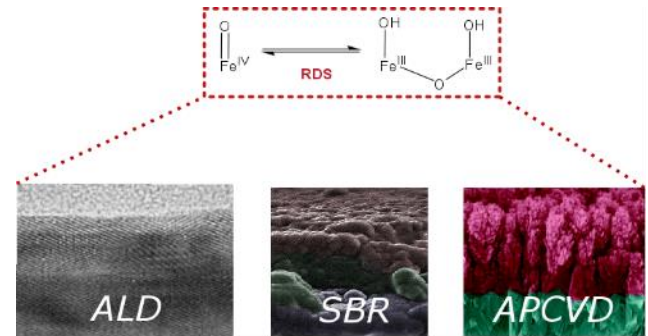
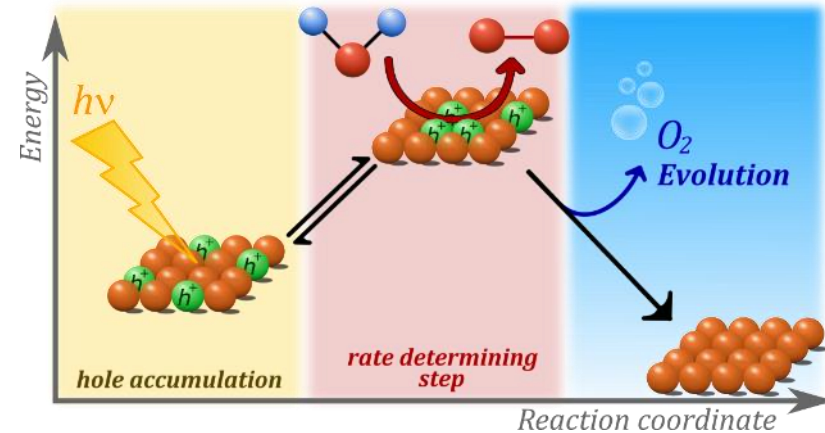
Final concluding remark

The studied materials ($\alpha\text{-Fe}_2\text{O}_3$, TiO_2 , BiVO_4 and WO_3) share common oxidative pathways of reaction, based on accumulation of charges, with kinetic differences only associated to the hole redox power

Differently synthesized hematite photoanodes perform water oxidation with the same mechanism

If you can't defeat them, ...

Change them for alcohol



Acknowledgements

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Dr. Yimeng Ma

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Reisner group – University of Cambridge

Batista group – Yale University



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